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A Suzuki cross-coupling route to substituted aziridines

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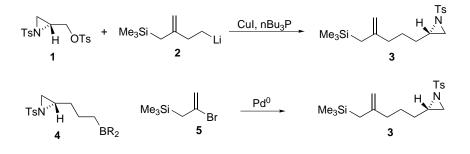
Abstract—We have shown that the Suzuki cross-coupling reaction of olefinic aziridines is an effective route for the synthesis of substituted aziridines. This is the first example of a palladium coupling reaction applied to an aziridine-containing molecule. This method is complementary to other methods of aziridine synthesis utilizing organocuprate reagents. © 2001 Elsevier Science Ltd. All rights reserved.

The Suzuki cross-coupling reaction provides an excellent method for coupling sp^3 hybridized carbons to sp^2 hybridized carbons (as well as sp^2 to sp^2). This reaction has been applied to the synthesis of a large variety of compounds.¹ Suzuki or similar types of reactions have not been carried out with an aziridine ring present in the molecule.^{2,3} Such a process in the presence of an aziridine ring suggests a number of potential problems, such as reaction of the organometallic portion (e.g. 4) with the aziridine ring,⁴ palladium-catalyzed aziridine ring opening,⁵ and hydrolysis of the aziridine ring under the basic reaction conditions (Scheme 1).

We were interested in preparing aziridine–allylsilanes such as **3** using our previously reported method involving the reaction of an organocuprate reagent (e.g. derived from **2**) with the aziridine **1**.⁶ Unfortunately, the synthesis of the requisite organolithium reagent proved troublesome.⁷ We thus began to explore the possibility of using a Suzuki coupling between the appropriately substituted aziridine (e.g. **4**) and an sp^2 bromide (e.g. **5**) to prepare the desired aziridines (**3**). This strategy could be very useful for the preparation of substituted aziridines, especially those in which the desired organometallic reagent (such as 2) is not readily available or one in which the aziridine is not amenable to the type of ring opening reaction we have typically employed.⁶

The Suzuki cross-coupling reaction between an aliphatic partner and a sp^2 -halide typically involves the conversion of the aliphatic partner to an organoborane via hydroboration. The appropriate olefin for hydroboration would be aziridine **9**. Olefin **9** was conveniently prepared from commercially available allylglycine (**6**). Treatment of amino acid **6** with MeOH/HCl followed by toluenesulfonyl chloride provided the diprotected amino acid **7**⁸ in 80% yield from allylglycine. The ester was then reduced with LiBH₄ followed by a Mitsunobu ring closure to yield the target aziridine **9** (Scheme 2).⁹

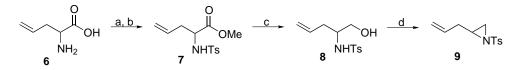
Olefin 9 was then treated with 9-BBN followed by 2-bromopropene (11) and $PdCl_2(dppf) \cdot CH_2Cl_2$ under Suzuki cross-coupling conditions. We were pleased to



Scheme 1.

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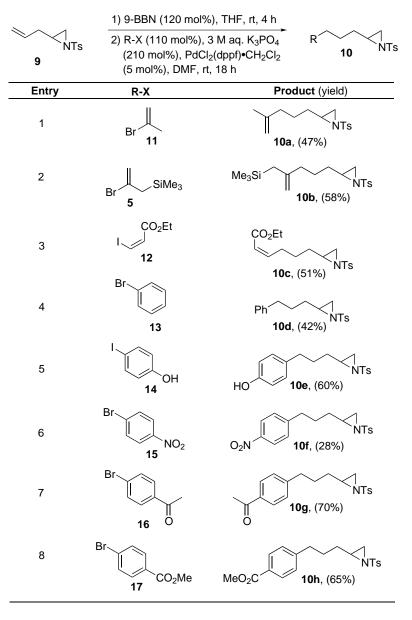
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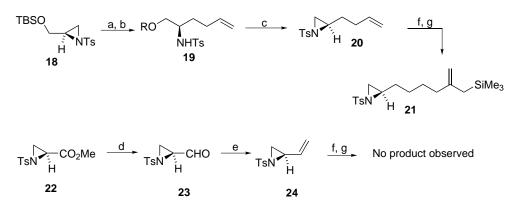


Scheme 2. (a) MeOH, AcCl (500 mol%), reflux, 24 h. (b) TsCl (110 mol%), Et₃N (250 mol%), CH₂Cl₂, rt, 24 h, 80% (two steps). (c) NaBH₄ (300 mol%), LiCl (300 mol%), EtOH:THF, 2:1, rt, 24 h, 80% (d) Ph₃P (110 mol%), DEAD (110 mol%), THF, rt, 16 h, 82%.

find that the coupling provided the desired coupling product in 47% yield (Table 1, entry 1). We have examined the coupling of **9** with a variety of vinylic and aryl halides (Table 1). The yields reported in Table 1 are of analytically pure material.⁹ As hoped, the bromoallylsilane **5** was an excellent cross-coupling partner providing **10b** in 58% yield. Similar yields were obtained from a number of other aryl halides (**13**, **14**, **16**, **17**). The only exception being 4-nitro-bromobenzene, which afforded **10f** in only 28% yield. Some of the low yields observed in the cross-coupling reactions can be attributed to difficulties in purification. All of the cross-coupling products (to some extent) co-eluted with 9-BBN by-products from the reaction and required additional purification. We should note that for many of these examples the organolithium reagent that might be required to prepare aziridines **10a**-h from aziridine **1** would not be readily accessible. Thus, this method is an excellent complement to the addition of an organometallic reagent to an aziridine as a route to substituted aziridines (Scheme 3).

Table 1. Examples of Suzuki cross-coupling reactions of olefinic aziridines9,10





Scheme 3. (a) AllylMgCl (300 mol%), CuCN (50 mol%), Et₂O:THF (1:1.5), -78 to 0°C, 45 h, R = Si(Me₂)*t*Bu, 75%. (b) *n*Bu₄NF (110 mol%), THF, 0°C, 1 h, R = H, 88%. (c) Ph₃P (110 mol%), DEAD (110 mol%), THF, 0°C to rt, 4 h, 93%. (d) DIBAL-H (160 mol%), CH₂Cl₂, -78° C, 3 h. (e) Ph₃PCH₂ (250 mol%), THF, -20° C, 1 h, 29% (two steps). (f) 9-BBN (120 mol%), THF, rt, 4 h. (g) 5 (110 mol%), 3 M aq. K₃PO₄ (210 mol%), PdCl₂(dppf)·CH₂Cl₂ (5 mol%), DMF, rt, 18 h, 21 (65%).

In an effort to extend this synthesis we prepared olefins 20 and 24. Olefin 20 was readily prepared from the serine derived aziridine 18.6a Reaction of 18 with allylmagnesium chloride/CuCN followed by removal of the silyl protecting group and Mitsunobu ring closure provided aziridine 20^{11} in excellent overall yield. Standard cross-coupling conditions using bromoallylsilane 5 provided substituted aziridine 21 in 65% yield.^{9,10} We next turned our attention to olefin 24. The known aziridine ester 22^{6a,12} can be reduced to aldehyde 23 by reaction with DIBAL-H. Wittig reaction of the aldehyde produces the vinyl aziridine 2413 in 29% yield over two steps. Unfortunately, hydroboration of 24, followed by Suzuki coupling conditions did not provide any of the desired product. We have examined a number of variations of the Suzuki cross-coupling protocol and all were unsuccessful. No aziridine containing products were obtained, and we believe that the aziridinyl borane formed upon hydroboration of **24** is unstable.

In conclusion, we have shown that the Suzuki coupling reaction of olefinic aziridines proceeds quite well. This reaction should be useful for the preparation of a variety of substituted aziridines.

Acknowledgements

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- 9. All new products showed satisfactory ¹H, ¹³C NMR and HRMS.
- 10. Typical procedure for the Suzuki cross-coupling reaction: The desired aziridinyl alkene (9 or 20, 1.1 mmol) was dissolved in dry THF (3.7 mL) and cooled to 0°C. A solution of 9-BBN (2.5 mL of a 0.5 M solution in THF) was added and the reaction was warmed to room temperature and stirred until no alkene remained (TLC, 20% EtOAc:hexanes, ca. 4 h). Distilled DMF (1.8 mL) was added followed by careful addition (H₂ evolution) of degassed aq. K₃PO₄ (3 M, 210 mol%). This was followed by a rapid addition of R-X (110 mol%, X=Br or I) and PdCl₂(dppf)·CH₂Cl₂ (5 mol%). The reaction was stirred overnight at room temperature. The solvent was removed in vacuo and the remaining DMF residue was partitioned

between Et_2O and saturated aq. NaHCO₃. The layers were separated and the aqueous layer was re-extracted with Et_2O . The combined organic layers were dried (MgSO₄), filtered, and concentrated in vacuo. The crude product was purified by column chromatography on silica gel.

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